

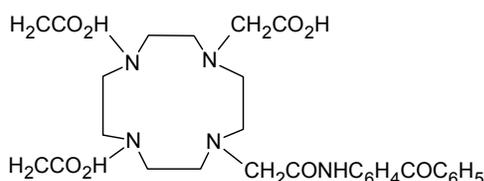
## Time-Gated Separation of Lanthanide Luminescence

## Introduction

The distinctive luminescence of lanthanide ions has long been a challenge for the spectroscopist. Methods to counter the extremely low absorption coefficients and luminescence quantum yields have concentrated on

- Protective ligands to reduce vibrational quenching effects by X–H moieties (X = O, N, ...).
- Deuterated solvents (also to reduce quenching effects of X–H).
- Antenna chromophores for enhanced absorption (with later energy-transfer to, and luminescence by, the lanthanide).

Luminescence from these systems comprises a number of sharp and well-resolved bands with characteristically long lifetimes, roughly  $10^{-6}$ – $10^{-3}$  s, depending upon the degree of quenching. With an integrated phosphorimeter, the Spex® FluoroMax®-P allows convenient time-gated measurement of lanthanide systems. In this *Application Note*, two lanthanide complexes comprising an encapsulating complex with benzophenone antenna chromophore (L, see Fig. 1) and europium (Eu) or terbium (Tb) ions are investigated using time-gated spectroscopy with the FluoroMax®-P.

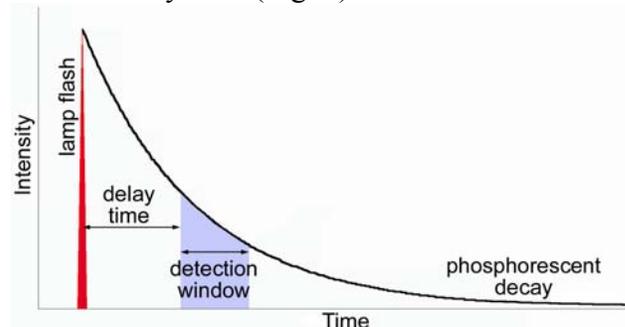


**Figure 1.** Benzophenone antenna chromophore, designated L in the text, complexed with Eu and Tb.

## Experiment &amp; Results

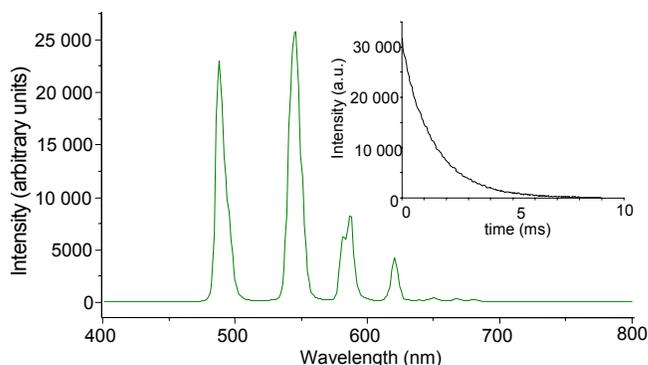
An excitation source (xenon lamp) flashes ( $\sim 3 \mu\text{s}$  pulse-width). Detection of the sample's luminescence occurs within a time-window after the flash, according to a chosen delay time and detection-window size. Acquisition of a spectrum corresponding to any desired time-slice during the luminescence decay is possible. Luminescence lifetime measurements are made by

recording the intensity of a specific wavelength with a chosen detection-window size and incremented delay time (Fig. 2).



**Figure 2.** Schematic showing how time-gated measurements are performed.

The individual spectra of Tb-L and Eu-L complexes are shown in Figs. 3 and 4. The resulting luminescence spectrum from a mixture of Tb-L and Eu-L (Fig. 5) combines features from both Tb-L and Eu-L, acquired with a delay time of 0.01 ms, and sample window of 10 ms. Thus results from detection of *all* the luminescence from *both* species appear in Fig. 5. Time-gating allows a mixture of the two species to be spectrally separated, based on their different lifetimes ( $\sim 0.60$  ms for Eu-L, and  $\sim 1.1$  ms for Tb-L).



**Figure 3.** Luminescence spectrum and decay for Tb-L(aq).  $\lambda_{\text{ex}} = 295$  nm;  $\lambda_{\text{em}} = 545$  nm.

Careful choice of gating parameters allows spectra from this mixture to be obtained selectively from an early time (to minimize Tb-L and maximize Eu-L) and a late time (with negligible Eu-L, approximating pure Tb-L), as shown in Fig. 6. A time-gated matrix scan (Fig. 7) shows

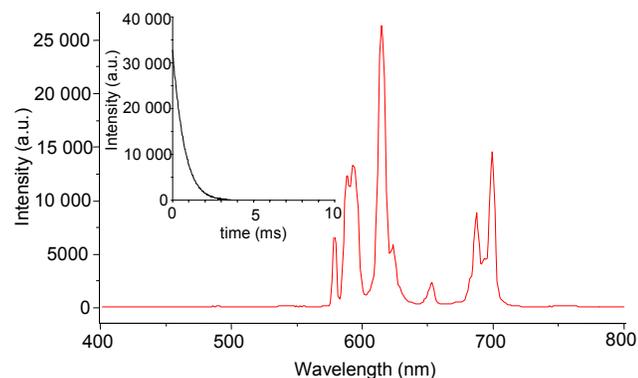
the dependence of the spectral profile with time. A number of complete emission spectra were recorded while automatically incrementing the delay time between each run. The result is a 3-D array, with intensity, wavelength, and time information.

### Conclusions

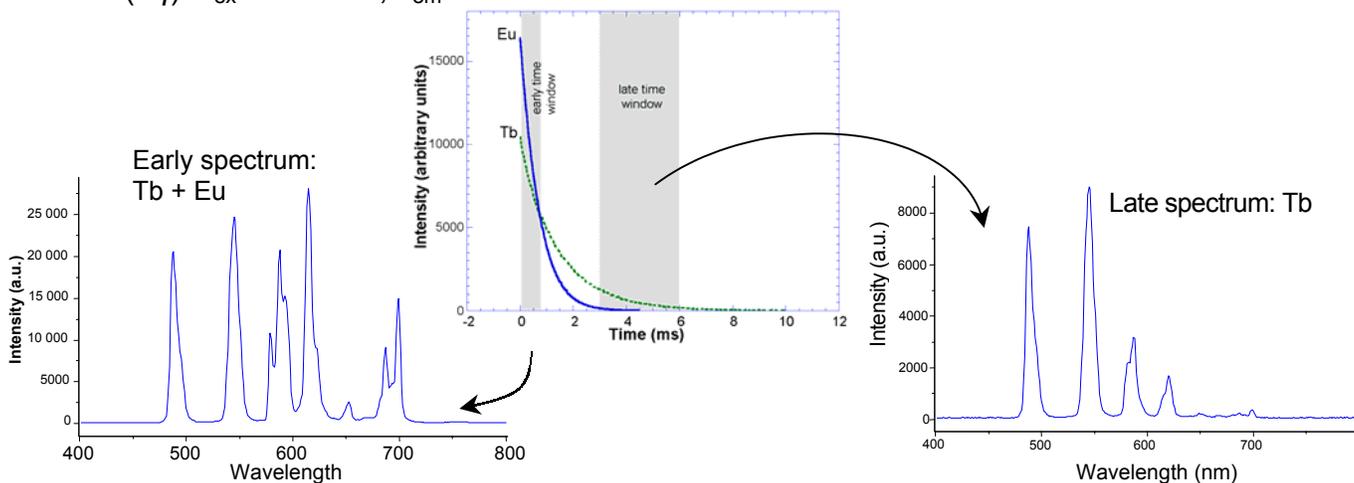
The Spex® FluoroMax®-P spectrofluorometer with integral flash-lamp can perform time-gated spectroscopic separation of mixtures of chemical species.

### Acknowledgements

We thank Drs. Andrew Beeby and Gareth Williams of the Chemistry Department, University of Durham, UK, for useful discussion and preparation of the compounds studied herein.



**Figure 4.** Luminescence spectrum and decay for Eu-L(aq).  $\lambda_{ex} = 295$  nm;  $\lambda_{em} = 614$  nm.



**Figure 6.** The principle of time-gated spectral separation for the mixture of Tb-L and Eu-L. The contribution from Eu-L is negligible in the spectrum recorded with the late time window.

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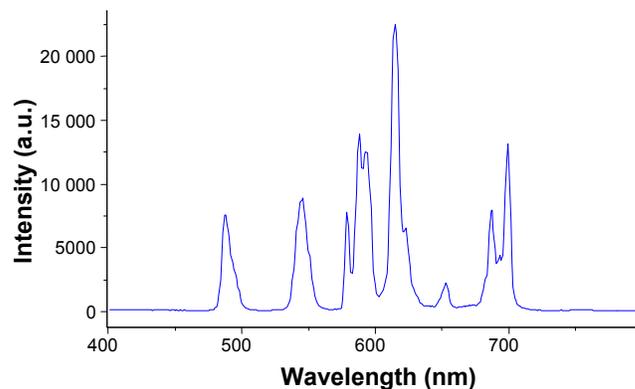
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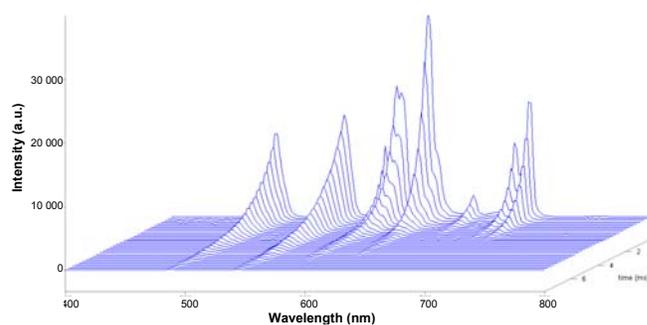
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**Figure 5.** Luminescence spectrum for mixed aqueous Tb-L and Eu-L.  $\lambda_{ex} = 295$  nm, delay = 0.010 ms, and sample window = 10 ms.



**Figure 7.** Time-gated matrix scan showing wavelength- and time-dependence of mixed aqueous Tb-L and Eu-L luminescence.